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FILING DATE FIRST NAMED INVENTOR APPLICATION NO. ATTORNEY DOCKET NO. CONFIRMATION NO. 09/750,125 12/29/2000 Q62454 6746 Mitsuhiro Kanada EXAMINER 01/21/2005 SUGHRUE, MION, ZINN, MACPEAK & SEAS, PLLC CHANG, VICTOR S 2100 Pennsylvania Avenue, N.W. Washington, DC 20037-3213 ART UNIT PAPER NUMBER 1771

DATE MAILED: 01/21/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)	
Office Action Summary	09/750,125	KANADA ET AL.	
	Examiner	Art Unit	
	Victor S Chang	1771	
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	correspondence addre	ess
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a reply - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	86(a). In no event, however, may a reply be ting within the statutory minimum of thirty (30) day rill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	nely filed s will be considered timely. the mailing date of this comm D (35 U.S.C. § 133).	nunication.
Status			
1) Responsive to communication(s) filed on 13 De	ecember 2004 and 13 October 20	<u>004</u> .	
2a) ☐ This action is FINAL . 2b) ☒ This	action is non-final.		
3) Since this application is in condition for allowar	•		erits is
closed in accordance with the practice under E	x parte Quayle, 1935 C.D. 11, 45	53 O.G. 213.	
Disposition of Claims			
4) Claim(s) 1-10,16 and 17 is/are pending in the a 4a) Of the above claim(s) is/are withdraw 5) Claim(s) is/are allowed.			
6) Claim(s) <u>1-10,16 and 17</u> is/are rejected.			
7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/or	coloction requirement		
	election requirement.		
Application Papers			
9) The specification is objected to by the Examiner			
10) The drawing(s) filed on is/are: a) acce			•
Applicant may not request that any objection to the one of the correction and the correction are the corrections.		` '	1 101/4)
11) The oath or declaration is objected to by the Ex			
Priority under 35 U.S.C. § 119			
 12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority documents 2. Certified copies of the priority documents 3. Copies of the certified copies of the priority 	s have been received. s have been received in Applicati	on No	
application from the International Bureau		ea in this National Sta	age
* See the attached detailed Office action for a list of	' ''	d.	
Attachment(s)	_		
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948)	4)		
2)	5) Notice of Informal P 6) Other:		52)

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DETAILED ACTION

Introduction

- 1. The Examiner has carefully considered Applicants' amendments and remarks filed on 12/13/2004 and 10/13/2004.
- 2. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.
- 3. Rejections not maintained are withdrawn.

Rejections Based on Prior Art

4. Claims 1-10 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 99/47573 either individually, or in view of admitted prior art, and further in view of Nakae et al. (US 4353817), generally as set forth in section 4 of Office action dated 4/13/2004, together with the following response to argument.

First, the Examiner repeats (see Office action dated 6/21/2002) the relied upon prior art references as follows: WO '573 is directed at a low-density microcellular thermoplastic elastomeric foams with closed cells. The foam is made using supercritical fluid CO₂ as the blowing agent (claims 1, 5-7) (Abstract). The polymer and the blowing agent are mixed in the melt stage in a tandem extruder under high temperature and pressure, subsequently the temperature and pressure are reduced to initiate foaming (page 3, lines 9-17). Alternatively, the polymers and any optional fillers are first melt compounded in a Brabender or twin-screw extruder, then mixed with the blowing agent

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for foaming (claims 3-4) (page 3, lines 18-25). Various foam properties such as the density, cell structure and size, compression set, etc. may be adjusted by varying the foaming conditions (page 5, lines 23-26), and it is noted that it is well known that CO₂ typically reaches supercritical fluid state under high pressure over 10 Mpa (claims 7-11). Suitable thermoplastic elastomers include blends of polyolefins, polyurethanes, etc. (claim 12) (page 2, lines 23-28). It should be noted that polyolefin and polyurethanes are inherently non-elastomeric thermoplastic polymer and thermoplastic elastomers, respectively. Finally, Applicants seems to admit the prior art JP-A-322168 teaches the method of impregnating a pre-formed unexpanded thermoplastic molding (claim 2) (Specification, page 4, paragraph 2). As to the flame retardant hydrated metal compound, it is noted that Nakae's invention is related to polymer foams with high flame retardancy (Abstract). Nakaen teaches that by adding hydrated metal compounds render the polymer foams highly flame retardant (column 2, lines 52 to column 3, line 11). Further, Nakae discloses that the hydrated metal oxide has a general structural formula M_mO_n, XH₂O, for example, aluminum hydroxide (Al₂O₃·3H₂O or Al(OH)₃). magnesium hydroxide (MgO·H₂O or Mg(OH)₂), etc. (column 7, lines 21-37). In the absence of unexpected results, it would have been obvious to one of ordinary skill in the art to modify the thermoplastic foams of WO '573 with the hydrated metal compounds taught by Nakae, including composite (mixture) metal hydrides, such as MgO·ZnO·H₂O, because it is prima facie to combine two compositions each of which is taught by prior art to be useful for the same purpose, so as to form a third composition to be used for

the very same purpose, motivated by the desire to improve the flame retardancy of the foams made by the methods taught by WO '573.

With respect to Applicants' argument "even if skilled artisan were to combine Nakae and WO '573, the combination would not result in the claimed invention" (Remarks, pages 3-4, bridging paragraph), the Examiner notes that Applicants fail to provide a clear reasoning why the combination would not result in the claimed invention. The Examiner reasserts that, in the absence of unexpected results, it would have been obvious to one of ordinary skill in the art to modify the thermoplastic foams of WO '573 with the hydrated metal compounds taught by Nakae, including composite (mixture) metal hydrides, such as MgO·ZnO·H₂O, because it is *prima facie* to combine two compositions each of which is taught by prior art to be useful for the same purpose, as set forth above.

Regarding Mr. Yamamoto's Declaration, filed 10/13/2004, Applicants' argument "the Declaration ... sets out clear evidence of unexpectedly superior results ... the use of Al₂O₃·nH₂O in Comparison Examples 1-3 resulted in low degrees of expansion and unacceptable flame retardancy ... the unique effect (unexpectedly high degree of expansion) of the presently claimed invention ... would not be obvious to one skilled in the art ..." (Remarks, page 4) has been carefully considered, but is not persuasive.

First, the Examiner notes that the Declaration appears to contain an error in identifying the metal hydride used in Experiment 4 as MgO·ZiO·H₂O (page 3), whereas as the table in page 4 correctly identifies the instantly claimed MgO·ZnO·H₂O, as recited in claim 1.

Second, the Examiner notes that Applicants' comparison of the degrees of expansion of

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thermoplastic foams containing Al₂O₃·nH₂O vs. MgO·ZnO·H₂O fail to recognize that it would have been obvious to one of ordinary skill in the art to use composite (mixture) metal hydrides, such as MgO·ZnO·H₂O, as set forth above. In particular, there is no data showing the degree of expansion of a relevant component metal hydride, such as MgO·H₂O, as taught by Nakae. Further, it is noted that Applicants' declaration states that the results are attributed to the affinity between a resin and a flame retardant (Declaration, page 5, first paragraph), the Examiner notes that such a statement also appears to indicate that the expansion results are expected whenever a MgO·H₂O (taught by Nakae) flame retardant is selected.

Conclusion

5. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Victor S Chang whose telephone number is 571-272-1474. The examiner can normally be reached on 8:30 - 5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Terrel H Morris can be reached on 571-272-1478. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

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Victor S Chang

Examiner

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1/14/2005